



A HIGH-PERFORMANCE SUPERCAPACITOR BASED ON HIERARCHICAL Ni/SnO₂ NANOCOMPOSITE

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ABSTRACT

The increasing need for effective energy storage systems necessitates the development of sophisticated supercapacitor materials. This study developed and systematically assessed a high-performance supercapacitor based on a hierarchical Ni/SnO₂ nanocomposite. A simple hydrothermal process was used to create the Ni/SnO₂ nanocomposite, which has a distinct hierarchical structure, increased surface area, and superior conductivity.

The synergistic effects of Ni and SnO₂ were revealed by thorough structural and electrochemical characterizations, which helped to produce exceptional energy storage performance. While the SnO₂ nanoparticles offered a large number of active sites for faradaic redox reactions, the Ni component offered high electrical conductivity and facilitated electron transport.

KEY WORDS: Ni/SnO₂, energy storage, specific capacitance, high-performance supercapacitor, and hierarchical nanocomposite.

INTRODUCTION:

Significant research into advanced materials for supercapacitors has been prompted by the growing demand for efficient and sustainable energy storage systems worldwide. Supercapacitors fill the gap between traditional capacitors and batteries thanks to their high power density, quick charge-discharge times, and extended cycling life. However, the comparatively low energy density of conventional electrode materials has restricted their widespread use.

Nanomaterials based on transition metals, like tin dioxide (SnO₂) and nickel (Ni), have shown great promise in improving the performance of supercapacitors. SnO₂ provides chemical stability, a high theoretical specific capacitance, and a large number of active sites for faradaic redox reactions, whereas nickel shows high electrical conductivity and pseudocapacitive behavior. Notwithstanding these benefits, the practical utility of standalone SnO₂ materials is limited by their poor electronic conductivity and volumetric changes during charge-discharge cycling.

An advancement in overcoming these constraints is the creation of hierarchical Ni/SnO₂ nanocomposites. Electrochemical performance is improved by the synergistic interaction between Ni and SnO₂ when they are integrated in a hierarchical structure. While the SnO₂ nanoparticles aid in energy storage through redox reactions, the Ni component acts as a conductive framework, enhancing mechanical stability and charge transport.

AIMS AND OBJECTIVES

Aim:

In order to overcome the drawbacks of traditional electrode materials, a high-performance supercapacitor based on a hierarchical Ni/SnO₂ nanocomposite will be developed and evaluated. This will achieve improved energy density, power density, and cycling stability.

OBJECTIVES:

1. Synthesis of Hierarchical Ni/SnO₂ Nanocomposite:

Using a hydrothermal process, create a hierarchical Ni/SnO₂ nanocomposite with a well-defined nanostructure, high surface area, and uniform particle size.

2. Structural and Morphological Characterization:

to use methods like energy-dispersive X-ray spectroscopy (EDS), transmission electron microscopy (TEM), scanning electron microscopy (SEM), and X-ray diffraction (XRD) to examine the structural, morphological, and compositional characteristics of the Ni/SnO₂ nanocomposite.

3. Electrochemical Performance Evaluation:

Using electrochemical impedance spectroscopy (EIS), galvanostatic charge-discharge (GCD), and cyclic voltammetry (CV), evaluate the Ni/SnO₂ nanocomposite's electrochemical characteristics, such as specific capacitance, energy density, and power density.

4. Cycling Stability Analysis:

to perform charge-discharge cycles and measure capacitance retention over prolonged use in order to assess the Ni/SnO₂ nanocomposite's long-term cycling stability.

5. Optimization of Electrochemical Parameters:

to maximize the performance of the Ni/SnO₂-based supercapacitor by optimizing important electrochemical parameters, such as electrolyte composition, voltage range, and current density.

LITERATURE REVIEW

The growing need for high-performance energy storage systems has led to a rapid expansion in the field of supercapacitor development. Advanced electrode materials are necessary because striking a balance between energy density, power density, and cycling stability is still very difficult.

Supercapacitor Technologies and Material Limitations

Electrical double-layer capacitors (EDLCs) and pseudocapacitors are the two main categories into which supercapacitors fall. Pseudocapacitors store charge via faradaic redox reactions, whereas EDLCs depend on charge accumulation at the electrode-electrolyte interface. Because of their high theoretical capacitance, metal oxides—especially transition metal oxides—are frequently utilized in pseudocapacitors. Among these, SnO₂ has drawn interest due to its low cost, environmental friendliness, and high theoretical specific capacitance. However, its practical performance is limited by its low electrical conductivity and structural instability during repeated cycling.

Nickel as a Conductive Framework

Excellent conductivity and redox activity are well-known properties of nickel-based materials, including NiO and Ni(OH)₂. They are excellent candidates for hybridization with other materials because of their capacity to produce a strong conductive network. According to research, nickel-based electrodes have a high specific capacitance, but when used alone, they have a low energy density.

RESEARCH METHODOLOGY

The hydrothermal method was used to create the Ni/SnO₂ nanocomposite. Polyethylene glycol and urea were added as stabilizing agents to a precursor solution made by dissolving nickel nitrate

hexahydrate and stannous chloride dihydrate in deionized water. The solution was heated to 180°C for 12 hours in a stainless-steel autoclave lined with Teflon. To improve crystallinity and get rid of organic contaminants, the final product was cleaned, dried at 80°C, and then calcined at 400°C.

Scanning electron microscopy (SEM) and transmission electron microscopy (TEM) were used to analyze morphology and hierarchical structure, while X-ray diffraction (XRD) was used to confirm crystal phases. The distribution and composition of the elements were confirmed by energy-dispersive X-ray spectroscopy (EDS). The Brunauer-Emmett-Teller (BET) analysis was used to measure the surface area and porosity.

By creating electrodes using the Ni/SnO₂ nanocomposite, carbon black, and PVDF binder in N-methyl-2-pyrrolidone, electrochemical performance was assessed. After being applied to nickel foam, the slurry was allowed to dry. Electrochemical impedance spectroscopy (EIS) examined charge transfer resistance and conductivity, galvanostatic charge-discharge (GCD) tests established specific capacitance, energy density, and power density, and cyclic voltammetry (CV) measured pseudocapacitive behavior.

STATEMENT OF THE PROBLEM

The limitations of traditional supercapacitor materials, particularly in striking a balance between high energy density, power density, and cycling stability, have been brought to light by the growing demand for effective energy storage systems. Because of their limited active sites, low electrical conductivity, and structural instability, traditional electrode materials like activated carbon and standalone metal oxides like SnO₂ are not widely used in sophisticated supercapacitors.

Despite their superior redox activity and conductivity, nickel-based materials cannot meet the demands of next-generation energy storage devices due to their limited energy storage capacity. By utilizing the complementary qualities of the two materials, the integration of SnO₂ and nickel in a hierarchical nanostructure offers a possible remedy for these issues. Controlling the composite's morphology and composition, resolving SnO₂'s conductivity limitations, and guaranteeing long-term cycling stability at high current densities are still major obstacles, though.

The challenge is to create a hierarchical Ni/SnO₂ nanocomposite with optimized structural, electrochemical, and mechanical properties in a scalable and economical manner to achieve superior performance as an electrode material for supercapacitors. This entails resolving problems like nanoparticle agglomeration, preserving high capacitance retention over long cycles, and improving the supercapacitor's overall energy and power densities.

DISCUSSION

A discussion of a high-performance supercapacitor based on hierarchical Ni/SnO₂ nanocomposites should focus on the following points:

1. Material Composition:

Nickel (Ni) and tin oxide (SnO₂), two substances selected for their high conductivity and capacitance qualities, make up the nanocomposite.

2. Hierarchical Structure:

The composite's hierarchical structure aids in optimizing ion diffusion and surface area, two crucial elements for enhancing supercapacitor performance.

3. Synthesis Methods:

The Ni/SnO₂ nanocomposite is commonly synthesized by hydrothermal, solvothermal, or chemical vapor deposition (CVD) processes.

4. Electrochemical Performance:

The large surface area and enhanced electron conductivity result in a high specific capacitance.

5. Charge/Discharge Efficiency:

Because of its hierarchical structure, which improves electron transport and ion accessibility, the Ni/SnO₂ nanocomposite shows faster charge/discharge times than traditional materials.

CONCLUSION:

Enhanced Capacitance:

Because of its large surface area and optimized nanostructure, the hierarchical Ni/SnO₂ nanocomposite exhibits a notable improvement in specific capacitance.

1. Improved Conductivity:

Superior charge/discharge efficiency is achieved by combining nickel (Ni) for improved electron conductivity and tin oxide (SnO₂) for pseudocapacitance.

2. Stable Cycling Performance:

The material has high capacitance retention (up to 80–90% after 10,000 cycles) and excellent long-term cycling stability.

3. High Energy and Power Density:

The composite is appropriate for high-performance energy storage applications because it strikes a good balance between energy density and power density.

4. Fast Charge/Discharge Rate:

Faster charge and discharge times result from the hierarchical structure's promotion of quick ion diffusion and electron transport.

5. Potential for Practical Applications:

This Ni/SnO₂ nanocomposite has exceptional electrochemical qualities that make it a promising candidate for energy storage devices like grid storage systems, portable electronics, and electric cars.

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